

Structural properties of the F4 MIL-140A(Ce) MOF by solid-state NMR spectroscopy



Andrea Giovanelli,^a Francesca Martini,^{a,b,c} Elisa Carignani, ^c Lucia Calucci,^{c,b} Marco Taddei,^{a,b} Marco Geppi^{a,b,c}

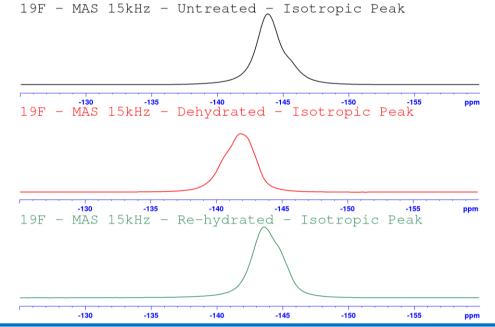
^aDipartimento di Chimica e Chimica Industriale, Università di Pisa, 56124 Pisa, Italy; ^bCentro per l'Integrazione della Strumentazione Scientifica dell'Università di Pisa (CISUP), 56126 Pisa, Italy; cIstituto di Chimica dei Composti Organo Metallici, Consiglio Nazionale delle Ricerche (CNR-ICCOM), 56124 Pisa, Italy

Introduction

Metal-Organic Frameworks (MOFs) are a class of crystalline compounds whose scaffolding derives from metal clusters or ions that are interconnected by organic linkers. Properties of MOFs are highly tunable and thus they can be employed in a wide range of fields of application. Solid State NMR has been employed to gain an in-depth knowledge of a MOF belonging to the MIL class, precisely F4 MIL 140A(Ce).

¹⁹F MAS Spectra

¹⁹F chemical shift is dependent on the presence of water. 19F - MAS 15kHz - Untreated - Isotropic Peak

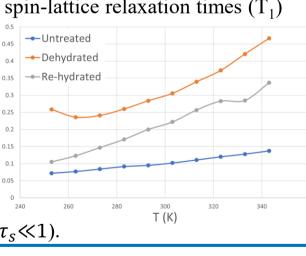


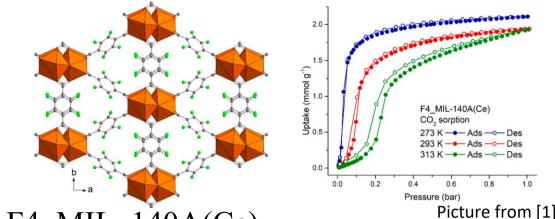
¹⁹F T₁ Relaxation Times

A dynamic process involving the TFBDC ligands has been detected by the trend of ¹⁹F spin-lattice relaxation times (T_1)

(s)

vs temperature. In the dehydrated sample the dynamic process is slower, and relaxation times are close to a minimum ($\omega_L \tau_s \simeq 1$), whilst in the other two samples dynamics is in the fast-motion regime ($\omega_L \tau_s \ll 1$).





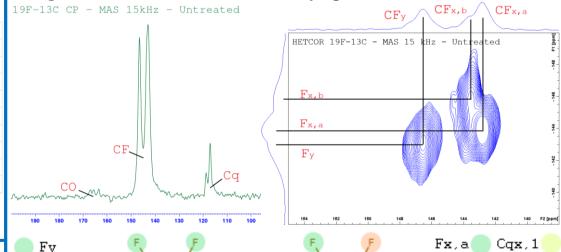
F4 MIL 140A(Ce)

F4 MIL 140A(Ce) is a relatively new MOF in which inorganic building units of Cerium^{IV} are interconnected by tetrafluoroterephtalates (TFBDC) ligands. This MOF is extremely promising for possible applications, in particular as a sorbent for gas separation, because of its water-based synthesis and its step-shaped CO₂ adsorption isotherm [1]. The Solid State NMR study has been carried out on three different samples:

- Untreated;
- After a dehydration process;
- After a partial re-hydration.

¹³C MAS Spectra

Using a combination of ¹⁹F-¹³C CP/MAS (Cross Polarization/Magic Angle Spinning) and HETCOR (HETeronuclear CORrelation) experiments, two different chemical environments in the solid have been observed. This is in agreement with data from X-Ray spectra.



Experimentals

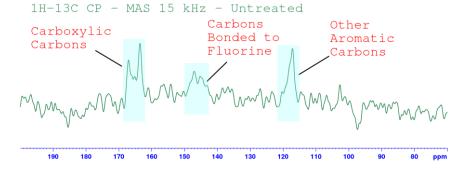
All spectra were recorded on a 3-channel Bruker Advance NEO 500 Solid State NMR spectrometer using a 4mm CP/MAS probe. Frequency Switched Lee Goldburg (FSLG) has been used in HETCOR experiments to remove homonuclear dipolar coupling. ¹H-¹³C and ¹⁹F-¹³C heteronuclear dipolar couplings have been always removed by a SPINAL-64 scheme. ¹⁹F T₁'s were measured by Inversion-Recovery.

Acknowledgements

CISUP (Centre for Instrument Sharing – University of Pisa) is acknowledged for the use of the Bruker Advance Neo 500 Solid State NMR spectrometer)



Moreover ¹H-¹³C CP has been employed to show that water molecules in the sample mainly interact with carboxylic moieties.



References: [1] ACS Sustainable Chem. Eng. 2019, 7, 1, 394–402